

Indian Institute of Technology Madras

Industrial Consultancy & Sponsored Research (IC&SR)

METHOD FOR MAKING NANOMETER THIN SHEETS OF METALS IN AIR

IITM Technology Available for Licensing

Problem Statement

- Molecular interactions at liquid-air interfaces were studied from Agnes Pockels times.
- Assembled structures at these interfaces have contributed to understanding **2D films**, even with diverse materials like nanoparticles.
- · Stable molecules and particles arrange at interfaces due to surface tension. The direct creation of nanostructures is possible at interfaces using atomic precursors.
- Methodologies for metal nanoparticles (NPs) on solid surfaces with ambient electrolytic spray and electrospray is adapted for liquid surfaces, allowing synthesis & assembly.
- · An electrical double layer at the liquid-air interface, along with its response to moderate **electric fields,** can induce motion in both surface and bulk liquid, guiding suspended NPs into organized arrangements.
- Traditional methods to create thin metal sheets in vacuum conditions use thermal and electron beam evaporation, and magnetron sputtering, but require high vacuum, sophisticated tools, and elevated temperature.
- Thus, there is a need for a new method that can address the above mentioned issues. The present invention introduces an ambient method for creating thin metal films.

Technology Category / Market

Technologies: Chemistry & Chemical Analysis, Micro & Nano Technologies

Application: Catalysis & Catalytic Conversion, Photovoltaics, Gas Adsorption, Energy Storage and Energy conversion, Electrochemistry, Coating.

Market: The global sheet metal market size was valued at US\$ 262.2 B in 2019 and is expected to grow by 5% CAGR from 2020 to 2025.

Industry: Chemical/Catalysts Manufacturing, Chemical Analysis, Instrumentation, Automotive and Aerospace, Electronics, Nanotech, Pharmaceutical and Biotechnology, Nanoelectronics Technology, Environmental Technology.

Technology

The present patents core idea is producing nanometer-thin metal films liquids on without vacuum or high temperatures.

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Key Features / Value Proposition

*<u>User Perspective:</u>

- Simple & Versatile: Easy to use method, allowing users to create nanometer-thin metal sheets with a variety of metal precursors & liquids as needed.
- Efficient Catalysis: The nanosheets serve as effective and reusable catalysts, improving efficiency of chemical reactions & promoting sustainability.
- Applications: The nanosheets Diverse find applications in catalysis, sensors, electronics, and energy storage, enabling users to explore a wide range of possibilities.

<u>*Technology Perspective:</u>

- Ambient Synthesis: The technology creates nanosheets without the need for high vacuum or extreme temperatures, simplifying the process and enhancing scalability.
- Material Flexibility & Controlled Self-Assembly: Nanosheets of different metals & alloys, are formed through precise electrospray-driven self-assembly, resulting in structured & consistent properties for various applications.

*Industrial Perspective:

- Industries benefit from a cost-efficient process, avoiding complex equipment & high-energy needs.
- Nanosheets enhance reaction rates & selectivity, valuable for industries reliant on catalysis.
- Market Diversification: Industries can expand product portfolios with nanosheets, exploring new markets and applications for innovation.

Fig 1 illustrates:

- a. schematic of electrospray deposition of PdCl₂ on water surface,
- b. mass spectrum of PdCl₂ solution in acetonitrile, using nESI source,
- c. optical image of electrospray deposition at air-water interface.



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Method

This is achieved by **depositing nanometer-scale droplets onto liquid surfaces under the influence of an electric field.**

With this objective, experiments are done by which **NPs of Pd**⁺ were synthesized on the surface of water reservoir that is self-assembled to form **nanoparticle-nanosheets**. **Fig 1, 2.**



Fig 2 illustrates: Mass spectrum collected after Pd NP-NS catalyzed coupling reaction between 4-tolylboronic acid and 4-bromophenol; inset shows the structures of the reactants and the product.

A method of making nanometer thin, <100 nm free standing 2D metal sheets at air-liquid interface - the said method comprises:

at least one metal salt precursor dissolved in a solvent;

•electrospraying of the solvated metal salt precursor at a voltage 1000-2000 V from 10-15 mm distance over the liquid surface;

 gently depositing nanometer scale droplets over the liquid surface at room temperature, which are then self- assembled to form nanoparticlenanosheets. Wherein, the charged metal ion reduced to neutral metal atom without using any external reducing agents during the course of deposition, that aggregates to form nanoparticles at the interface and arrange to form nanoparticlenanosheets over the liquid surface.

Wherein

- The metal is palladium and the metal salt is palladium chloride.
- The **metals** are selected from various **salts** of **gold**, **silver**, **platinum**, **palladium**, **nickel and copper**.
- The thin film is made of different alloys including silver-palladium and gold-palladium.
- The **solvent** of the **precursor solution** includes **methanol**, **ethanol**, **water**, **acetonitrile and combination** thereof with different proportions.
- The liquid reservoir contains liquids including water, ionic liquid and ethylene glycol.
- The **liquid reservoir** contains **liquid metals or semiconductors** which are solids at room temperature.
- The electrospray deposition of films occurs along with other stimuli including light, temperature and magnetic field.
- The modification of the electrospray occurs during solvent evaporation of the droplets using temperature, light, electric & magnetic fields.
- The thin metal films are used as a catalyst, sensor, gas adsorbing medium and electrodes for electrochemical reactions.

Intellectual Property

IITM IDF Number: 1594 IP Patent Number: 332623 (Granted) PCT/IN2018/050654

TRL (Technology Readiness Level)

TRL – 4; Technology validation in Lab

Research Lab

Prof. Pradeep T Department of Chemistry

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