

IIT MADRAS Technology Transfer Office TTO - IPM Cell



Industrial Consultancy & Sponsored Research (IC&SR)

# PROCESS FOR PREPARING METAL FREE NITROGENOUS ORDERED MESOPOROUS CARBON MATERIAL AND ITS PRODUCT THEREOF **IITM Technology Available for Licensing**

# Problem Statement & Unmet Need

Indian Institute of Technology Madras

- > The commercial application of alkaline fuel cells is determined by the efficiency of oxygen reduction reaction(ORR)and on catalysts that promote these reactions
- > Currently used platinum catalysts have sluggish kinetics, methanol crossover, high cost and limited availability.

## Intellectual Property

- IITM IDF Ref. 2122
- IN410837(Granted)

## Technology Category/ Market

#### Category – Manufacturing / Chemical

Applications –Catalysis. separation and Energy Storage, Drug Delivery Systems, Diagnostics Industry- Nanotechnology, Chemicals, Manufacturing, Medical

**Market** -The global microporous and mesoporous materials market size was USD 8,601.7 million in 2021. The market is projected to touch USD 14.930 million by 2028 exhibiting a CAGR of 8.2% during the forecast period.

# Key Features / Value Proposition

#### Technical Perspective:

- □ Nitrogenous ordered mesoporous carbon as catalyst for oxygen reduction reaction in alkaline medium displaying superior activity and stability with a current density on par with 20% Pt/C commercial catalyst along with a four electron transfer and peroxide yield less than 10%.
- □ High surface area and uniform pore size

#### **User Perspective:**

- □ High efficiency, eco-friendly by product,costeffective stationary and portable applications and flexibility of usage in transportation
- □ These catalysts can be effectively used in fuel cells as cathodes for ORR to avoid cross over effects

### **CONTACT US**

Dr. Dara Ajay, Head Technology Transfer Office, IPM Cell- IC&SR, IIT Madras

IITM TTO Website: https://ipm.icsr.in/ipm/

## Technology

- The present invention discloses a process for preparing nitrogenous ordered mesoporous carbons using hexagonal or cubic silica templates and nitrogen-containing organics
- Further discloses metal free nitrogenous ordered mesoporous carbon material using the hard template method taking polyvinyl pyrrolidone as a precursor thereby avoiding metal chlorides as polymerizing agents and studied their ORR abilities in alkaline medium for fuel cell
- Said nitrogenous ordered mesoporous carbon as . catalyst for oxygen reduction reaction in alkaline fuel cells

#### **METHOD:**



Email: smipm-icsr@icsrpis.iitm.ac.in sm-marketing@imail.iitm.ac.in Phone: +91-44-2257 9756/ 9719



IIT MADRAS Technology Transfer Office TTO - IPM Cell



Industrial Consultancy & Sponsored Research (IC&SR)

# **IITM Technology Available for Licensing**

Image



Indian Institute of Technology Madras

Figure 1 illustrate TEM images of: (a) MNC-359 and (b) MNC-859

- The hexagonal or cubic silica templates are selected from SBA-15 and KIT-6
- The template free carbon product, wherein synthesised ordered mesoporous nitrogen containing carbon were denoted as (a) MNC-359 and MNC-859 respectively



Figure 2 illustrate [A] Hydrodynamic linear sweep voltammograms in oxygen saturated 0.1 N KOH electrolyte at a rotation speed of 1600 rpm with a scan rate of 20 mV/s and [B] Peroxide yields and electron numbers calculated from transfer the RRDE measurement results of prepared 25 metal-free Ndoped ordered mesoporous carbons

# **CONTACT US**

Dr. Dara Ajay, Head Technology Transfer Office, IPM Cell- IC&SR, IIT Madras

IITM TTO Website: https://ipm.icsr.in/ipm/ Email: smipm-icsr@icsrpis.iitm.ac.in sm-marketing@imail.iitm.ac.in Phone: +91-44-2257 9756/ 9719



Figure 3 illustrate [A] Chronoamperometric stability responses of N-doped ordered mesoporous carbons at 0.7 V from the oxygen saturated 0.1 N KOH for the catalysts. [B] Methanol tolerance responses of N-doped ordered mesoporous carbons compared to commercial Pt/C catalyst at 0.7 30 V from the oxygen saturated 0.1 N KOH for the catalysts

### TRL (Technology Readiness Level)

### TRL-4 Technology validated in lab

# Research Lab

Prof. SELVAM P NCCR & Dept. of Chemistry, IIT Madras