



### AN INTEGRATED CDI ELECTRODE IITM Technology Available for Licensing

#### Problem Statement

- In the present era, the availability of clean potable water at affordable cost is a growing challenge for mankind. To meet the growing demand of drinking water, many materials & methods are used, but having difficulties due to **expensive** process/implementation, & **consumption of high energy**.
- There are many conventional methods discussed herein does not provide solutions as discussed in the present invention. Hence, there is a need to address above issues.

#### Technology Category/Market

**Technology:** Integrated CDI Electrode;

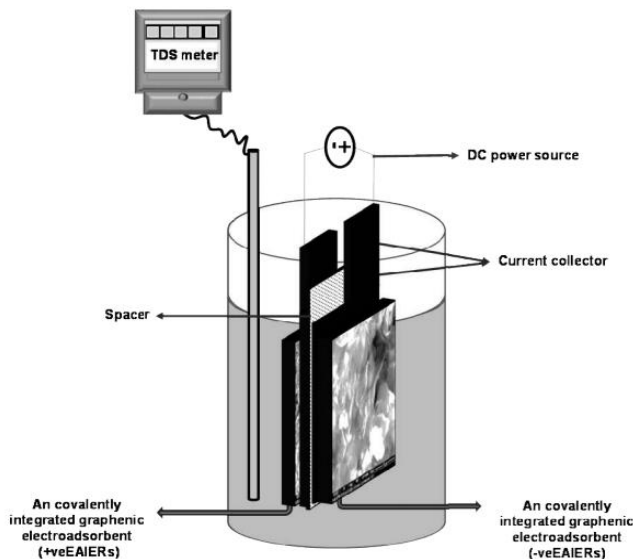
**Industry:** Chemical Industry, Grey Water Management Industries; **Application:** Environment Engineering, Waste water treatment;

**Market:** The global **capacitive deionization** market is projected at a **CAGR** of **5.4%** during (2023-2026)

#### Technology

- Present invention describes the preparation of **an electrode assembly** for **capacitive deionization (CDI)** comprising:
- **One electrode** made of chemically-linked reduced graphene **oxide-cation** exchange resins molecular constructs synthesized by in-situ polymerization of styrene & **reduced graphene oxide (RGO)** followed by functionalization of reduced graphene oxide@polystyrene(**RGO-PS**) composite with sulfonate and/or carboxylate moiety.
- **One electrode** made of chemically-linked reduced graphene **oxide-anion** exchange resins molecular constructs synthesized by in-situ polymerization of styrene & **reduced graphene oxide (RGO)** followed by functionalization of reduced graphene oxide@polystyrene(**RGO-PS**) composite with amine moiety.

- The electro-desorption during CDI does not lead to re-adsorption & consequent reduction of adsorption capacity of the overall system.
- The covalent construct is mixed with additives includes carbon nanotubes (CNTs), fullerenes and carbon fibers to enhance the performance.
- The electrodes are used for removing salts from blackish water.



**Fig. 1** Illustrates representation of a **CDI set-up** used for measuring **CDI performance**.

#### Intellectual Property

**IITM IDF Ref. 1637; Patent No:345270;**  
**PCT Application No. PCT/IN2018/050894**  
**US Application No. US 16/958,971**

#### TRL (Technology Readiness Level)

**TRL-4**, Proof of concept tested in Lab;

#### Research Lab

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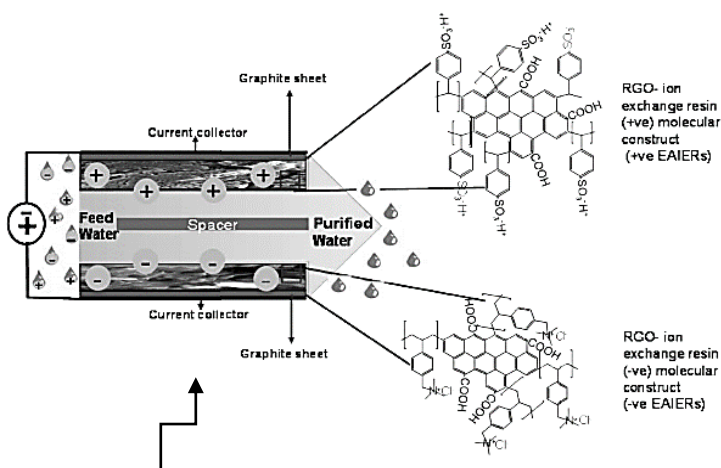
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### Key Features / Value Proposition

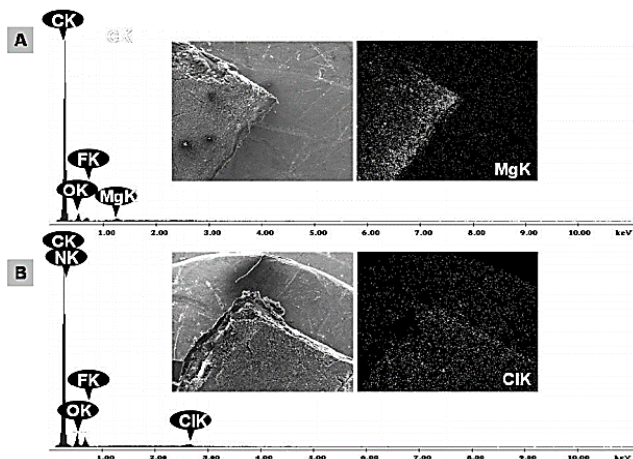
#### ❖ **Technical Perspective & Industrial Perspective:**

- ❖ **Types of Resins used:** The resin selected from the group of polystyrene, polysulphone, poly-acrylic acid, polyacrylamide, poly(methylmethacrylate), polyimide, polyethylene terephthalate, polyethylene grafted maleic anhydride, polypropylene, poly(vinyl alcohol), poly(vinyl chloride) & polyethylene vinyl acetate & combination thereof.
- ❖ **Types of Electrode material/ two-dimensional materials used:** The electrode material is selected from the group of reduced graphene oxide, molybdenum disulfide(MoS<sub>2</sub>), tungsten diselenide(WSe<sub>2</sub>), tungsten disulfide(WS<sub>2</sub>) and combination thereof.
- ❖ **Removing Cations & Anions and positive-negative charges:** The electrodes remove  
→ cation of different charge(including Fe<sup>3+</sup>, Mg<sup>2+</sup>and Na<sup>+</sup>),  
→anions of different charge (including Cl<sup>-</sup>,NO<sub>3</sub><sup>-</sup>, F<sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, arsenite & arsenate ions,  
→positive & negative charged ions are removed from impure water.
- ❖ **Efficiency:** The assembly is used along with other deionization & purification methods to enhance removal efficiency.
- ❖ **Cost-effective, provides single solution & applicable in the domestic/Industrialist area.**

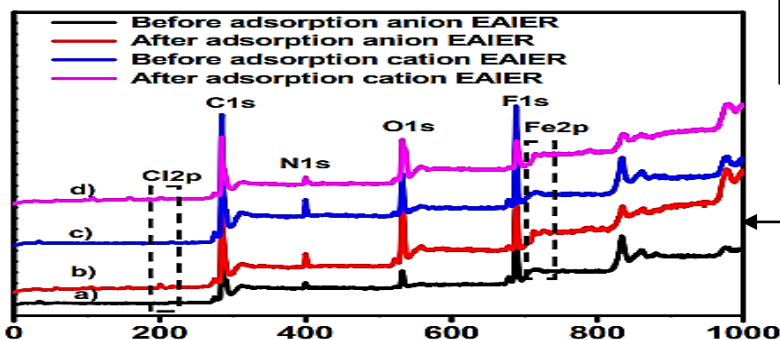
### Image



**Fig. 2: Illustrates schematic design of a cell with integrated EAIERs (molecular construct) for capacitive deionization (CDI)**



**Fig. 3 (above): Illustrates SEM EDS of MgCl<sub>2</sub> adsorption after a single adsorption cycle on A) cathode and B)anode, & the corresponding SEM EDS & elemental mapping images shown**



**Fig. 4 (Left) : Illustrates XPS survey spectra of the material after single adsorption cycle, (a) anode and (c) cathode (before adsorption); (b) anode and (d) cathode (after FeCl<sub>3</sub> adsorption).**

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